Chemistry. - "The Crystalforms of Some Organic Nitrogen-compounds. By Prof. F. M. Jaeger.
(Communicated at the meeting of January 30, 1926).
§ 1. In this paper the crystalforms are described of the following fifteen nitrogen-derivatives, which have been investigated during the last few years: p-Amino-acetophenone; Acetyl-phenyl-urethane; Acetyl-$\alpha$-amino-pyridine; $\alpha$-Amino-pyridine-urethane; Acetyl-chinine; Tartrona-minic-acid; d-Tartramidic-acid; Trihydrazide-hydrochloride of Tricarballylic acid; Amido-sulphonic acid; Phtalimidine : Ammonium-phtalimidineacetate; Ethylenediamine-hydrochloride; p-Nitro-mono-propyl-aniline; Benzoylcyano-acetic methyl-ether and m-Nitro-benzoylcyano-acetic methylether. Also the description is added here of Triethylphosphine-carbodisulphide, the remarkable addition-product of carbondisulphide: $C S_{2}$ and $P\left(C_{2} H_{5}\right)_{3}$.
§ 2. p-Amino-Acetophenone ${ }^{1}$ ): $\mathrm{CH}_{3} . \mathrm{CO} . \mathrm{C}_{6} \mathrm{H}_{4}\left(\mathrm{NH}_{2}\right) ; \mathrm{Mpt}: 106^{\circ} \mathrm{C}$.
From benzene in large, flat, pale yellowish, transparent crystals; from a mixture of alcool + chloroform in rectangular plates, elongated either parallel to the $b$-, or to the a-axis (fig. 1).


Fig. 1.
Monoclinic-prismatic.

$$
\begin{gathered}
a: b: c=1,7447: 1: 1,6662 \\
\beta=73^{\circ} 16^{\prime}
\end{gathered}
$$

Forms observed: $c=\{001\}$, mostly predominant and lustrous; $a=\{100\}$, well developed; $m=\{110\}$, broader than $p=\{120\} ; r=\{\overline{1} 01\}$, well developed and yielding good reflections. The habit is often elongated parallel to the $b$-axis, with $\{001\}$ and $\{100\}$ predominant, and in the last case laso prismatic along the c-axis.
${ }^{1}$ ) V. B. Drewsen, Lieb. Ann. der Chemie, 212, 162, (1882).

Angular values: Observed: Calculated:

$$
\begin{aligned}
& a: r=(100):(\overline{1} 01)={ }^{\star} 55^{\circ} \quad 9^{\prime} \\
& a: m=(100):(110)={ }^{*} 596 \\
& c: a=(001):(100)={ }^{*} 7316 \\
& c: r=(001):(\overline{1} 01)=5133 \\
& p: p=(120):(\overline{120})=3355 \\
& p: m=(120):(110)=143 \\
& c: p=(001):(120)=85 \quad 20 \frac{1}{2} \\
& c: m=(001):(110)=8135 \\
& \text { - } \\
& 51^{\circ} 35^{\prime} \\
& 3319 \\
& 14 \text { 141 } \\
& 85 \quad 16 \\
& 81 \quad 29
\end{aligned}
$$

In the prism-zone the angular values vary not inappreciably. No distinct cleavability. The optical axial plane is $\{010\}$; the first bisectrix is almost perpendicular to $\{001\}$. The angle of the optical axes is evidently very small; weak and inclined dispersion.
§ 3. Acetyl-phenyl-Urethane ${ }^{1}$ ): $\mathrm{C}_{6} \mathrm{H}_{5} \cdot \mathrm{~N}\left\langle_{\mathrm{CO}}^{\mathrm{CO}} . \mathrm{CH}_{3} \mathrm{OC}_{2} \mathrm{H}_{5}\right.$; Mit.: $59^{\circ} \mathrm{C}$.
From ligroine in colourless, well-developed crystals. (Fig. 2).
Rhombic-bipyramidal. $a: b: c=1,2323: 1: 15141$.

Forms observed: $c=\{001\}$, large and lustrous; $m=\{110\}$, also well developed and giving sharp images; $a=\{100\}$ and $b=\{010\}$, generally almost equally large; $o=\{211\}$, well reflecting, mostly smaller, but occasionally much larger than $m$. The habit is short-prismatic with tabular form parallel to $\{001\}$, or that of thick plates parallel to $\{001\}$.

Angular values:


Fig. 2.
Observed: Calculated:


No distinct cleavage could be found.

[^0]The plane of the optical axes is probably $\{100\}$; the c-axis is first bisectrix. The apparent axial angle is only small; the dispersion is weak.
§ 4. Acetyl- $\alpha$-Amino-pyridine ${ }^{1}$ ): $\mathrm{C}_{5} \mathrm{H}_{4}$ N.NH. $\mathrm{CO} . \mathrm{CH}_{3}: M p t: 71^{\circ} \mathrm{C}$. From benzene + ligroine in colourless, very lustrous crystals. (Fig. 3).

> Monoclinic-prismatic.

$$
a: b: c=1,4939: 1: 2,0719 ;
$$

$$
\beta=89^{\circ} 14^{1} / 2^{\prime} .
$$



Fig. 3.
Forms observed: $c=\{001\}$, predominant, yielding good reflections; $r=\{100\}$ and $s=\{101\}$, equally broad, well developed and giving sharp images; $q=\{011\}$, broad and lustrous. The habit is that of thick tables parallel to $\{001\}$, with elongation in the direction of the b-axis.

Angular values: Observed: Calculated:

$$
\begin{array}{lll}
c: r & =(001):(101)=^{*} 53^{\circ} & 42 \frac{1}{2}^{\prime} \\
c: s=(001):(\overline{101})=^{*} 54 & 42 \frac{1}{2} & - \\
c: q=(001):(011)==^{*} 64 & 14 & - \\
r: s=(101):(10 \overline{1})=71 & 35 & 71^{\circ} 35^{\prime} \\
q: q=(011):(01 \overline{1})=51 & 32 & 51 \\
r: q 2 \\
r: q=(101):(011)=74 & 59 & 75
\end{array}
$$

Perfectly cleavable parallel to $\{010\}$.
§ 5. a-Amino-pyridine-Urethane ${ }^{2}$ ): $\mathrm{C}_{5} \mathrm{H}_{4} \mathrm{~N} . \mathrm{NH} . \mathrm{CO} . \mathrm{OC}_{2} \mathrm{H}_{5} ; \mathrm{Mpt}$ : $105^{\circ}, 5 \mathrm{C}$.

From ether this substance is obtained in well developed, but very soft and plastic crystals; they are colourless and commonly possess a prismatic or parallelopipedal shape. By the translation-planes present the crystals can only be manipulated, if the necessary precautions be taken; the angular values vary not inappreciably in the different zones.

Monoclinic-prismatic.

$$
\begin{gathered}
a: b: c:=0,7946: 1: 1,3552 ; \\
\beta=73^{\circ} 21^{\prime} .
\end{gathered}
$$

[^1]Forms observed: $m=\{110\}$, large and lustrous; $a=\{100\}$, narrower than $m$, often absent, but giving good reflections; $c=\{100\}$, well devel-


Fig. 4.
oped and yielding very good reflections; $r=\{101\}$, well reflecting, much broader than $s=\{108\}$, which form is often absent.

Angular values: Observed: Calculated:

$$
\begin{aligned}
& m: m=(110):(1 \overline{10})={ }^{\star} 74^{\circ} 34 \\
& c: m=(001):(110)={ }^{*} 76 \quad 49 \frac{1}{2} \\
& c: r=(001):(101)=4740 \\
& c: s=(001):(108)=\begin{array}{lll}
10 & 40 & 10^{\circ} 54^{\prime}
\end{array} \\
& a: m=(100):(110)=37 \quad 23 \quad 3717 \\
& a: s=(100):(108)=63 \quad 0 \quad 62 \quad 27 \\
& r: s=(101):(108)=3645 \quad 3646
\end{aligned}
$$

Cleavage perfect parallel to $\{101\}$.
On $\{001\}$ diagonal, on $\{110\}$ oblique extinction.
§ 6. Acetyl-chinine: $\left.{ }^{1}\right) \mathrm{C}_{20} \mathrm{H}_{23}\left(\mathrm{CO} . \mathrm{CH}_{3}\right) \cdot \mathrm{N}_{2} \mathrm{O}_{2} ; \mathrm{Mpt}: 116^{\circ} \mathrm{C}$.
This compound crystallizes at room-temperature from ether in splendid, colourless and transparent crystals, having a high lustre and measuring often several cm . in length. (Fig. 5).

## Rhombic-bisphenoidal.

$$
a: b: c=1,1142: 1: 0,6119
$$

Forms observed: $a=\{100\}$ and $m=\{110\}$, about equally large and very lustrous; they yield excellent reflections, although on a often multiple images. Further: $b=\{010\}$, much narrower, often only with a single plane, well reflecting; $s=\{201\}$ and $r=\{101\}$, both giving excellent

[^2]images, $s$ mostly much smaller than $r ; o=\{\overline{1} 11\}$, large and lustrous, almost always showing a single plane; $\omega=\{111\}$, small, dull, but well ${ }^{1}$ measurable. The crystals are well built and have constant angular values. Their habit is thick-prismatic and elongated parallel to the c-axis or flattened parallel to two planes of $m$ and a.

.Fig 5.

Angular values: Observed: Calculated:


No distinct cleavage was found.
In solution the substance is levogyrate: $[a]_{D:}=-120,{ }^{\circ} 8$.
§ 7. Tartronaminic Acid ${ }^{1}$ ) : NH 2. $\mathrm{CO} \cdot \mathrm{CH}(\mathrm{OH}) . \mathrm{COOH}$; Met : $160^{\circ} \mathrm{C}$. with decomposition.

From water this compound crystallizes commonly in colourless flat needles, which are arranged in spherolithes, but rarely also in big, colourless, prismatic crystals. They are well built and allow accurate measurements.

Monoclinic-prismatic. $a: b: c=3,0003: 1: 2,4063$;

$$
\beta=64^{\circ} 19^{\prime}
$$

Forms observed: $\mathrm{c}=\{001\}$ and $a=\{100\}$, both giving very sharp images and almost equally broad; $o=\{111\}$,


Fig. 6.
splendidly reflecting and much better than $\omega=\{\overline{1} 11\}$, which form shows only narrow and dull faces. The habit is thick-prismatic parallel to the b-axis.

Angular values: Observed: Calculated:

$$
\begin{aligned}
& a: c=(100):(001)={ }^{*} 64^{\circ} 19^{\prime} \\
& c: o=(001):(111)={ }^{*} 5931 \\
& a: o=(100):(111)={ }^{*} 6216 \\
& \text { o:o }=(111):(\overline{1} 1 \overline{1})=7034 \quad 70^{\circ} 23^{\prime} \\
& c: \omega=(001):(\overline{1} 11)=74 \quad 8 \quad 74 \quad 4 \frac{1}{4} \\
& \omega: 0=(\overline{1} 11):(\overline{1} 1 \overline{1})=46 \quad 21 \quad 46 \quad 24 \frac{3}{4} \\
& a: \omega=\overline{(100)}:(\overline{1} 11)=80 \quad 59 \quad 81 \quad 9 \\
& \omega: 0=\overline{(111)}:(111)=3640 \quad 36
\end{aligned}
$$

No distinct cleavage could be observed.
On a and $c$ normal extinction. The plane of the optical axes is $\{010\}$; inclined dispersion. One branch of the hyperbola is on $\{001\}$ visible at the border of the field of the microscope.
§ 8. Dextrogyrate Tartramidic Acid ${ }^{1}$ ) : $\mathrm{COOH} . \mathrm{CHOH} . \mathrm{CHOH} . \mathrm{CO}$. $N H_{2}$; Mpt: $172^{\circ} \mathrm{C}$.

Crystals of the shape reproduced in Fig. 7 were obtained from a


Fig. 7. solution in water. They are colourless, highly lustrous and yield excellent images.

Rhombic-bisphenoidal; (pseudotetragonal).

$$
a: b: c:=0,7352: 1: 0,7393
$$

Forms observed: $m=\{110\}$. $r=\{201\}$ and $q=\{011\}$, all about equally well developed, very lustrous and giving good reflections; $\omega=\{221\}$, somewhat smaller, but yielding sharp images, as also $b=\{010\} ; c=\{001\}$, distinct, but a little duller. The crystals are nearly isometrical, with a slight flattening in the direction of the a-axis.

Angular values: Observed: Calculated:

$$
\begin{aligned}
& b: m=(010):(110)={ }^{\star} 53^{\circ} 40 \frac{2_{3}^{\prime}}{} \\
& q: q=(011):(0 \overline{1} \underline{1})={ }^{\star} 7257 \\
& r: r=(201):(20 \overline{1})=5258 \quad 52^{\circ} 52 \frac{1}{3} \\
& c: r=(001):(201)=63 \quad 31 \quad 63 \quad 33 \frac{9}{3} \\
& b: \omega=(010):(221)=5643 \quad 56 \text { 38 } \frac{1}{8}
\end{aligned}
$$

[^3]Angular values：Observed：Calculated：

$$
\begin{aligned}
& \omega: r=(221):(201)=3318 \quad 33 \quad 21 \frac{1}{2} \\
& m: m=(110):(1 \overline{10})=72 \quad 39 \frac{1}{2} \quad 72 \quad 38 \frac{2}{3} \\
& b: q=(010):(011)=53 \quad 31 \frac{1}{2} \quad 53 \quad 31 \frac{1}{2} \\
& c: q=(001):(011)=36 \quad 28 \frac{1}{2} \quad 36 \quad 28 \frac{1}{2} \\
& m: q=(110):(011)=49 \quad 37 \frac{1}{2} \quad 49 \quad 37 \frac{1}{2} \\
& r: q=(201):(011)=69 \quad 8 \quad 69 \quad 1 \\
& m: r=(110):(201)=4341 \quad 4349 \frac{2}{3}
\end{aligned}
$$

No distinct cleavage was found．
On $\{010\}$ and $\{001\}$ normal extinction．The plane of the optical axes is $\{100\}$ ，with the $c$－axis probably as first bisectrix．Pasteur observed in the crystals of＇the corresponding levogyrate isomeride，instead of $\omega$ ，a form $o=\{221\}$ ．
§9．Triethylphosphine－Carbodisulphide ${ }^{1}$ ）：$P\left(\mathrm{C}_{2} \mathrm{H}_{5}\right)_{3} . \mathrm{CS}_{2} ; \mathrm{Mpt}: 122^{\circ} \mathrm{C}$ ．
This compound crystallizes from alcohol in flat， blood－red，very brittle needles．

Monoclinic－prismatic（pseudo－rhombic）．

$$
\begin{gathered}
a: b: c=0,6306: 1: 0,9716 ; \\
\beta=89^{\circ} 511 / 2 .
\end{gathered}
$$

Forms observed：$b=\{010\}$ ，strongly predominant and，just as $r=\{101\}$ and $r^{\prime \prime}=\{\overline{10} 1\}$ ，yielding very sharp images；$\omega=\{111\}$ ，narrow，but well mea－ surable；$c=\{001\}$ ，lustrous and well reflecting．


Fig． 8.

Angular values：Observed：Calculated：

$$
\begin{aligned}
& r: r^{\prime \prime}=(101):(\overline{1} 01)={ }^{*} 114^{\circ} \quad 2^{\prime} \\
& b: \omega=(010):(111) \text { 二 }^{\star} 629 \\
& c: r=(001):(101) \text { 二 }^{\star} 5655 \\
& r: \omega=(101):(111)=2751 \\
& 27^{\circ} 51^{\prime} \\
& r: b=(101):(010)=90 \quad 0 \quad 90^{\circ}
\end{aligned}
$$

Perfectly cleavable parallel to $\{010\}$ and $\{101\}$ ．The plane of the optical axes is practically parallel to $\{001\}$ ；the first bisectrix is perpendicular to $\{010\}$ and has a positive character．Very strongly dichroitic：on $b$ for vibrations parallel to $\{001\}$ dark－red，for those perpendicular to this， orange－yellow．

[^4]§ 10. Trihydrazide-hydrochloride- of Tricarballylic Acid ${ }^{1}$ ): $\mathrm{C}_{3} \mathrm{H}_{5}(\mathrm{CO}$. $\left.\mathrm{NH} . \mathrm{NH}_{2}\right)_{3}, 3 \mathrm{HCl}$.
From alcool by slow evaporation in vacuo beautiful, colourless crystals were obtained.

## Cubic-hexakisoctahedral.

Well built octahedra, with (111): $(\overline{1} 1)=70^{\circ} 33^{\prime}$. The small corrosionfigures on the faces of $\{111\}$ were hexagons with their edges parallel to those of the octahedron. The crystals are isotropous, without circular polarisation.
§ 11. Amidosulphonic Acid: $\mathrm{NH}_{2} . \mathrm{SO}_{3} \mathrm{H}$.
From a solution in water big, colourless, splendidly built and perfectly transparent crystals are obtained, as reproduced in Fig. 9.

$$
\begin{gathered}
\text { Rhombic-bipyramidal; } \\
(\text { pseudo-tetragonal }) . \\
a: b: c=0,8720: 1: 0,8774 .
\end{gathered}
$$



Fig. 9

Forms observed: $c=\{001\}$, large
and very lustrous; $o=\{111\}$, also well developed and yielding sharp reflections; $r=\{102\}$ and $q=\{021\}$, almost equally large and well reflecting; $m=\{210\}$, appreciably smaller, but distinct and exactly measurable; $\omega=\{114\}$, very narrow and giving somewhat duller images. The habit is almost isometrical, with flattening parallel to $\{001\}$; often $t$ and $\omega$ show only half the number of faces, the crystals therefore showing an apparent monoclinic symmetry. However, the crystals are really pseudotetragonal: the angle (001): $(101)=45^{\circ} 10^{\prime} \frac{1}{2}$, so that the $b$-axis would be the vertical axis of the apparently tetragonal individual; a:c becomes then about: $1: 1,142$. The extinction on $\{210\}$ and the other optical properties prove, however, the true rhombic symmetry of the crystals.

Angular values: Observed: Calculated:

$$
\begin{aligned}
& \mathrm{c}: \mathrm{o}=(001):(111)={ }^{*} 53^{\circ} \quad 93^{\prime} \\
& m: m=(210):(\overline{210})={ }^{*} 47 \quad 7 \\
& c: q=(001):(021)=60 \quad 20 \frac{1}{4} \quad 60^{\circ} 19 \frac{1}{2}^{\prime} \\
& q: q=(021):(02 \overline{1})=59 \quad 19 \quad 59 \quad 21 \\
& c: r=(001):(1 \underline{10})=26 \quad 57 \frac{1}{2} \quad 26 \quad 42 \frac{1}{4} \\
& 0: 0=(111):(\overline{1} 1)=63 \quad 30 \quad 63 \quad 28 \frac{2}{3} \\
& o: o=(111):(11 \overline{1})=73 \quad 40 \frac{1}{2} \quad 73 \quad 40 \frac{1}{2} \\
& c: \omega=(00 \overline{1}):(11 \overline{2})-18 \quad 45 \quad 18 \quad 27
\end{aligned}
$$

No distinct cleavage could be found.
The plane of the optical axes is $\{001\}$, with the $b$-axis as first bisectrix. The apparent axial angle is large, the dispersion weak.
${ }^{1}$ ) Th. Curtius and A. Hesse, Journ. für prakt. Chem., 62, 236, (1900).

Evidently the crystals are identical with those described by A. Fоск (Zeits. f. Kryst., 14, 531, (1888)), where $\{010\}=\{001\}$ in our crystals.
§ 12. Phtalimidine: $\quad \mathrm{C}_{6} \mathrm{H}_{4}<\mathrm{CO}_{2}>\mathrm{NH} ; \mathrm{Mpt}: 150^{\circ} \mathrm{C}$.
From a mixture of chloroform and carbondisulphide flat parallelogram-shaped crystals were obtained, as reproduced in fig. 10.

Monoclinic-prismatic.

$$
\begin{gathered}
b: c=1: 0,4913 ; \\
\beta=79^{\circ} 20^{2} / 3^{\prime} .
\end{gathered}
$$

Forms observed: $b=\{010\}$, strongly predominant and yielding sharp reflections, as also $a=\{100\} ; q=\{100\}$, well developed.
Angular values : Observed: Calculated:

| $a: q=(100):(011)=^{*} 80^{\circ}$ | 25' |  | - |
| :---: | :---: | :---: | :---: |
| $b: q=(010):(011)={ }^{*} 64$ | 131 |  | - |
| $q: q=(011):(011)=51$ | $33 \frac{1}{4}$ | 51 |  |

$$
a: b=(100):(010)=90 \quad 0 \quad 90^{\circ}
$$



Fig. 10.

Perfectly cleavable parallel to $\{010\}$; distinctly parallel to $\{100\}$.
The plane of the optical axis is probably parallel to $\{010\}$. On $b$ one of the optical principal sections makes an angle of $191 / 2^{\circ}$ with the c-axis in the posterior quadrant.
§ 13. Ammonium - Phtalimidine - acetate :

$$
\mathrm{C}_{6} \mathrm{H}_{4}<\underset{\mathrm{CH}_{2}}{\mathrm{CO}}>\mathrm{N}-\mathrm{CH}_{2}
$$ $\mathrm{COO}\left(\mathrm{NH}_{4}\right)+2 \mathrm{H}_{2} \mathrm{O}$.

This salt was deposited from an aqueous solution in the form of big, colourless cristals of the shape reproduced in Fig. 11.


Fig. 11.

Monoclinic－prismatic．

$$
\begin{gathered}
a: b: c=0,8837: 1: 1,4301: \\
\beta=77^{\circ} 46^{\prime} .
\end{gathered}
$$

Forms observed： $\mathrm{c}=\{001\}$ ，strongly predominant and giving excellent images；$q=\{011\}$ ，broad，also very lustrous；$r=\{101\}$ ，broad reflecting； $a=\{100\}$ ，narrow and dull，but occasionally broader than $r$ and yielding good reflexes；$s=\{\overline{1} 01\}$ ，narrower than $r$ ，well measurable ；$\omega=\{122\}$ ， small，well reflecting；$o=\{\overline{3} 62\}$ ，larger than $\omega$ and yielding rather good images．The habit is tabular parallel to $\{001\}$ with elongation in the direction of the $b$－axis．

Angular values：Observed：Calculated：

No distinct cleavability．
§ 14．Ethylenediamine－hydrochloride： $\mathrm{C}_{2} \mathrm{H}_{4}\left(\mathrm{NH}_{2}\right)_{2}, 2 \mathrm{HCl}$ ．


Fig． 12.

From water at room－temperature，this compound is obtained in splendid，colourless，perfectly trans－ parent and strongly refracting crystals．They allow exact measurements to be made．（Fig．12）．

Monoclinic－prismatic．

$$
a: b: c=1,4496: 1: 0,6597
$$

$$
\beta=88^{\circ} 26^{3} / 4^{\prime} .
$$

Forms ${ }_{\alpha}^{\text {ob }}$ observed：$m=\{110\}$ ，very lustrous and yielding excellent images，as also $o=\{111\} ; c=\{001\}$ ． small，well measureable；$\omega=\{122\}$ ，in most cases only rudimentary，with strongly curved faces；often even totally absent．The habitus is that of flat，kite－ shaped crystals or of steep needles or long prisms．

$$
\begin{aligned}
& c: r=(001):(101)={ }^{*} 49^{\circ} 40 \\
& r: a=(101):(100)={ }^{*} 286 \\
& c: q=(001):(011)={ }^{*} 5425 \\
& a: s=(100):(10 \overline{1})=35 \quad 2 \quad 34^{\circ} 48^{\prime} \\
& s: c=(10 \overline{1}):(00 \overline{1})=67 \quad 27 \quad 67 \quad 26 \frac{1}{3} \\
& q: q=(011):(01 \overline{1})=71 \quad 12 \quad 71 \quad 10 \\
& r: q=(101):(011)=67 \quad 54 \quad 67 \text { 52⿺𠃊⿻丷木女⿱⿰㇒一乂 } \\
& s: q=(10 \overline{1}):(01 \overline{1})=76 \quad 55 \quad 77 \quad 6 \\
& a: q=(100):(011)=8248 \quad 8255 \\
& c: \omega=(001):(122)=54 \quad 4 \quad 53 \quad 53 \frac{1}{2} \\
& \omega: o=(122):(36 \overline{2})=30 \quad 30 \quad 30 \quad 21 \\
& o: c=(36 \overline{2}):(00 \overline{1})=95 \quad 46 \quad 95 \quad 45 \frac{1}{3} \\
& a: \omega=(100):(122)=58 \quad 41 \quad 59 \quad 5 \\
& \omega: q=(122):(011)=2353 \quad 2350
\end{aligned}
$$

Angular values: Observed: Calculated:

| $m: m=(110):(\overline{10} 0)=$ | $10^{\circ}$ | 47' | - |  |
| :---: | :---: | :---: | :---: | :---: |
| $c: m=(001) ~: ~(110) ~=* ~$ | 89 | 7 | - |  |
| o : o = (111) : $(1 \overline{1} 1)={ }^{*}$ | 61 | 26 |  |  |
| $m: 0=(1 \overline{10}):(111)=$ | 77 | 56 | $77^{\circ}$ | 581 ${ }^{\prime}$ |
| $m: 0=(110):(111)=$ | 50 | 40 | 50 | 451 |
| $\mathrm{c}: 0=(001):(111)=$ | 38 | 12 | 38 | $21 \frac{1}{2}$ |
| $m: \omega=\overline{1} 10):(122)=$ | 71 | 19 | 71 | 1 |

It seems that no distinct cleavage is present here.
§ 15. p-Nitro-mono-Propyl-Aniline: $\mathrm{C}_{6} \mathrm{H}_{4}\left(\mathrm{NO}_{2}\right) \cdot \mathrm{NH}\left(\mathrm{C}_{3} \mathrm{H}_{7}\right)$.
From alcohol this compound is deposited in the form of elongated, deep yellow, flat prisms with oblique terminal faces. They are highly lustrous, have a lilac surface-hue and yield multiple images. (Fig. 13).

## Triclinic-pinacoidal.

$$
\begin{aligned}
& a: b: c=0,6135: 1: 0,2992 \\
& A=59^{\circ} 38^{\prime} ; a=64^{\circ} 42^{\prime} . \\
& B=125 \\
& 21 ; \beta=121^{\circ} 17 \frac{1}{2}^{\prime} . \\
& C=72
\end{aligned} 38: \gamma=89^{\circ} 30^{\prime} .
$$

Forms observed: $b=\{010\}$, predominant and, like $a=\{100\}$ and $c=$ $=\{001\}$, all yielding very sharp reflexes; $m=\{230\}, p=\{130\}$ and $t=\{2 \overline{3} 0\}$, all about equally broad and well reflecting: $q=\{011\}$, distinctly developed, lustrous; $s=\{0 \overline{1} 1\}$, small and faintly reflecting. The habit is elongated in the direction of the $c$-axis, with flattening parallel $\{010\}$.

Angular values: Observed: Calculated:

| $c=(100):(001)=$ * |  | 39' | - |  |
| :---: | :---: | :---: | :---: | :---: |
| $a: b=(100) ~:(010)=$ * |  | 22 | - |  |
| $b: c=(010) ~: ~(001) ~=~$ |  | 22 | - |  |
| $c: q=(001):(011)={ }^{*}$ | 14 | 13 |  |  |
| $p: b=(130):(010)=^{*}$ | 33 | 31 | - |  |
| $b: t=(0 \overline{10}):(2 \overline{30})=$ | 39 | 22 | $39^{\circ}$ | 15' |
| $a: t=(100):(2 \overline{30})=$ | 33 | 16 | 33 | 23 |
| $a: m=(100):(230)=$ | 48 | 1 | 48 | 17 |
| $m: p=(230) ~: ~(130) ~=~$ | 25 | 40 | 25 | 34 |
| $b: q=(010):(011)=$ | 106 | 9 | 106 | 9 |
| $c: s=(001):(0 \overline{1} 1)=$ | 11 | 3 | 11 | 31 |
| $s: b=(\overline{1} 1):(0 \overline{10})=$ | 48 | 35 | 48 | 341 |
| $c: m=$ (001) : $(230)=$ | 82 | 42 | 82 | 54 |



Fig. 13.

On $\{010\}$ oblique extinction; the optical principal section makes, on
this face, an angle of about $12^{\circ}$ with the direction of the c-axis. Strongly dichroitic: orange-yellow-pale yellow.
§ 16. Methylether of Benzoyl-cyano-acetic Acid: (CN). $\mathrm{CH}(\mathrm{CO}$. . $\mathrm{C}_{6} \mathrm{H}_{5}$ ) $\mathrm{CO} . \mathrm{OCH}_{3}$; Mpt: $75^{\circ}, 2 \mathrm{C}$.

The compound was prepared from benzoylchloride and the sodiumsalt of methyl-cyano-acetate at a low temperature. ${ }^{1}$ )

From absolute alcohol the substance crystallizes at $60^{\circ} \mathrm{C}$ in beautiful, pale yellow, very lustrous prisms, which are often flattened parallel to two opposite planes of $m$.

## Rhombic-bipyramidal.

$$
a: b: c:=1,1433: 1: 0,7052
$$

Forms observed: $m=\{110\}$, predominant and yielding excellent reflections; $q=\{011\}$, large and very lustrous; $o=\{111\}$, smaller than


Fig. 14,
q. often apparently in two bisphenoids: the corrosion-figures on $m$ and o prove, however, beyond doubt that the symmetry is really bipyramidal. Further: $b=\{010\}$, very narrow, often hardly visible and yielding only weak images; $p=\{120\}$, well developed and lustrous, often absent; $r=\{101\}$, small and dull; $s=\{021\}$, also subordinate, but giving good reflections. The habit is short-prismatic parallel to the c-axis. A number of combinations were observed, dependent on circumstances during the crystallisation: e.g.. : $\{110\},\{011\}$ and $\{111\} ;\{110\},\{011\},\{111\},\{021\}$ and $\{010\} ;\{110\}$ and $\{111\} ;\{110\}\{111\}$ and $\{011\} ;\{110\},\{111\},\{011\},\{021\}$, $\{010\} ;\{101\}$ and $\{120\}$; etc. Compare Fig. 14.

[^5]Angular values: Observed: Calculated:

$$
\begin{array}{rlll}
o: o & =(111):(\overline{1} 1)={ }^{\star} 86^{\circ} & 16^{\prime} & - \\
m: q=(110):(011)={ }^{\star} 64 & 17 \frac{1}{2} & - \\
m: m=(110):(1 \overline{1} 0)=97 & 35^{\frac{1}{2}} & 97^{\circ} 39^{\prime} \\
q: q=(011):(0 \overline{1} 1)=70 & 18 & 70 & 23 \\
m: o=(1 \overline{1} 0):(111)=46 & 52 & 46 & 52 \\
o: o=(111):(1 \overline{1} 1)=61 & 54 & 61 & 57 \\
o: r=(111):(101)=30 & 57 & 30 & 58 \frac{1}{2} \\
b: p=(010):(120)=23 & 44 & 23 & 37 \frac{1}{2} \\
p: m=(120):(110)=17 & 32 & 17 & 33 \\
m: r=(110):(101)=69 & 39 \frac{1}{2} & 69 & 47 \\
r: q=(101):(011)=45 & 57 \frac{1}{2} & 45 & 55 \frac{1}{2} \\
b: s=(010):(021)=35 & 29 & 35 & 20 \\
s: q=(021):(011)=19 & 22 & 19 & 28 \frac{1}{2}
\end{array}
$$

No distinct cleavability was stated.
§ 17. Methyl-ether of $m$-Nitro-benzoyl-cyano-acetic Acid: $\mathrm{C}_{6} \mathrm{H}_{4}\left(\mathrm{NO}_{2}\right)$ $\mathrm{CO} . \mathrm{CH}(\mathrm{CN}) . \mathrm{CO} . \mathrm{OCH}_{3}$.

The substance was prepared in an analogous way as the former, $m$-Nitrobenzoylchloride being used in this case.

From a hot solution in ether or from a mixture of ether and acetone, small, almost colourless, very lustrous crystals were obtained.

## Monoclinic-prismatic.

$$
\begin{gathered}
a: b: c=0,7958: 1: 0,8067 \\
\beta=82^{\circ} 9^{\prime} .
\end{gathered}
$$

Forms observed: $a=\{100\}$ and $m=\{110\}$, well developed and highly lustrous ; $c=\{001\}$, also large, but much duller ; $r=\{905\}$ and $o=\{111\}$, distinct, but yielding mostly only faint reflections. The angles in the zone of the $b$-axis oscillate within rather wide limits with separate


Fig. 15
individuals. The habit is short-prismatic parallel the c-axis or tabular parallel $\{001\}$. See Fig. 15.

Angular values: Observed: Calculated:

$$
\begin{aligned}
& a: m=(100):(110)={ }^{*} 51^{\circ} 45^{\prime} \\
& m: 0=(110):(111)={ }^{*} 3249 \\
& m: c=(110):(001)={ }^{*} 85 \quad 9 \\
& m: m=(110):(1 \overline{10})=7630 \quad 76^{\circ} 30^{\prime} \\
& a: r=(100):(905)=\begin{array}{llll}
26 & 53 \frac{1}{2} & 26 & 48
\end{array} \\
& r: c=(905):(001)=55 \quad 10 \frac{1}{2} \quad 55 \quad 21 \\
& a: c=(100):(101)=82 \quad 4 \quad 82 \quad 9 \\
& c: o=(001):(111)=52 \quad 20 \quad 5220
\end{aligned}
$$

Distinctly cleavable parallel to $\{001\}$. The corrosion-figures obtained on $\{100\}$ with Cassia-oil, were in full agreement with the symmetry mentioned.

On a and $c$ normal extinction. The plane of the optical axes is $\{010\}$; the dispersion is inclined and extraordinarily strong, with: $\varrho>v$.

The apparent axial angle (in Cassia-oil) is about $47 \frac{1}{2}^{\circ}$; one of the axes is seen almost perpendicular to $\{001\}$, the first bisectrix almost halving the angle of the edges of (100) and (001) with (010).

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[^0]:    ${ }^{1}$ ) D. R. NijK, Rec. d. Tray. shim. d. Pays-Bas, 39, 701, (1920).

[^1]:    ${ }^{1}$ ) E. Dingemanse and J. P. Wibaut, Rec. d. Trav. Chim. d. Pays-Bas, 42, 240, (1923),
    ${ }^{2}$ ) E. Dingemanse and J. P. Wibaut, Rec. d. Trav. Chim. d. Pays-Bas, loco cit.

[^2]:    1) L. Seekles, 'Dissertation, Leiden, (1922), p. 33.
[^3]:    ${ }^{1}$ ) Comp, e. g.: L. Pasteur, Ann. de Chim. et Phys., (3), 38, 454, (1853); P. Groth, Chem. Kryst., III, 302, 308. The preparation here investigated was made by Dr.R.A.WEERMAN Diss. Delft, (1916), p. 109.

[^4]:    ${ }^{1}$ ）According to J．P．Wibaut（Rec．d．Trav．Chim．d．Pays－Bas，44，（1925），239）the meltingpoint of $95^{\circ} \mathrm{C}$ ．，as given in literature，must be considered as wrong．With some other choice of the parameters，the crystals investigated would be identical with those investigated by Sella（Rend．Acad．Lincei，（2），20，361，（1863）．

[^5]:    ${ }^{1}$ ) See also: L. Barthe, Compt. rend. Paris, 106, 1416, (1888) ; A. Haller, ibid., 101, 1270, (1883).

